

Iron 10 T 0.05 - 1 mg/l Fe Ferrozine / Thioglycolate

Instrument specific information

The test can be performed on the following devices. In addition, the required cuvette and the absorption range of the photometer are indicated.

Instrument Type	Cuvette	λ	Measuring Range
SpectroDirect, XD 7000, XD 7500	□ 10 mm	562 nm	0.05 - 1 mg/l Fe

Material

Required material (partly optional):

Reagents	Packaging Unit	Part Number
Iron II LR (Fe ²⁺)	Tablet / 100	515420BT
Iron II LR (Fe ²⁺)	Tablet / 250	515421BT
Iron LR (Fe ²⁺ und Fe ³⁺)	Tablet / 100	515370BT
Iron LR (Fe ²⁺ und Fe ³⁺)	Tablet / 250	515371BT

Application List

- Waste Water Treatment
- · Cooling Water
- Boiler Water
- Galvanization
- Drinking Water Treatment
- Raw Water Treatment
- Pool Water Treatment

Preperation

 Water that has been treated with organic compounds such as corrosion inhibitors, must be oxidised where necessary to break down the iron complex. 1 ml of concentrated Sulphuric acid (≥ 95 %) and 1 ml concentrated Nitric acid (≥ 65 %) is therefore added to to 100 ml water sample and boiled down to approximately half the volume. After cooling down, the digestion procedure is continued.

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Notes

- 1. This method is for the determination of total dissolved Fe^{2+} and Fe^{3+} .
- For the determination of Fe²⁺, the IRON (II) LR Tablet, instead of the IRON LR Tablet is used.
 Variations in the length of the vial can extend the measuring range:
- 10 mm vial: 0.05 mg/l 1 mg/l, solution: 0.01
- 20 mm vial: 0.025 mg/l 0.5 mg/l, solution: 0.01
- 50 mm vial: 0,.1 mg/l 0.2 mg/l, solution: 0.001

Digestion





10 min

Fill a suitable sample vessel Add 1 ml concentrated with 100 ml sample .

sulfuric acid (≥ 95 %).

The sample is to be heated for 10 minutes, or for as long as it takes for everything to be completely dissolved.







Allow the sample to cool to room temperature.

Adjust pH-value of the sam- Fill the sample with deiople with ammonia solution nised water to 100 ml. (10-25 %) to 3-5.

This sample is used for the analysis of total solved and dissolved Iron.

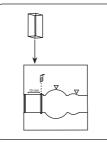
Implementation of the provision Iron (II,III), dissolved with Tablet

Select the method on the device

For testing of total solved and dissolved Iron, carry out the described digestion.

For this method, no ZERO measurements are to be carried out with the following devices: XD 7000, XD 7500 $\,$



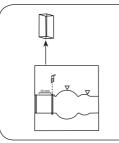




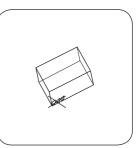
Fill 10 mm vial with sample.

Place **sample vial** in the sample chamber. • Pay attention to the positioning.

Press the ZERO button.



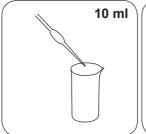




Remove **vial** from the sam- Empty vial. ple chamber.

Dry the vial thoroughly.

For devices that require no ZERO measurement, start here.



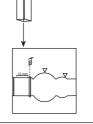




Fill a suitable sample vessel Add IRON LR tablet. with 10 ml sample .

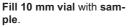
Crush tablet(s) by rotating slightly and dissolve.





Test

Press the **TEST** (XD: **START**) button.



Place **sample vial** in the sample chamber. • Pay attention to the positioning.

Wait for **5 minute(s) reac**tion time.

Once the reaction period is finished, the measurement takes place automatically.

The result in mg/l Iron appears on the display.

5 min

Chemical Method

Ferrozine / Thioglycolate

Appendix

Interferences

Removeable Interferences

 The presence of copper increases the test result by 10%. At a concentration of 10 mg/l copper in the sample, the measurement result is increased by 1 mg/l iron. The interference can be eliminated by the addition of thiourea

Bibliography

Photometrische Analyse, Lange/ Vjedelek, Verlag Chemie 1980, p. 102

^{a)} determination of free, combined and total | ^{b)} Reactor is necessary for COD (150 °C), TOC (120 °C) and total -chromium, - phosphate, -nitrogen, (100 °C) | ^a MultiDirect: Adapter is necessary for Vacu-vials[®] (Order code 19 20 75) | ^a Spectroquant[®] is a Merck KGaA Trademark | ^a alternative reagent, used instead of DPD No.1/No.3 in case of turbidity in the water sample caused by high concentration of calcium and/or high conductivity | ^a additionally required for determination of bromine, chlorine dioxide and ozone in the presence of chlorine | ^a Reagent recovers most insoluble iron oxides without digestion | ^b additionally required for samples with hardness values above 300 mg/l CaCO₃ | ^a high range by dilution | ^a including stirring rod, 10 cm