Name of manufacturer: KANTO CHEMICAL CO., INC.

Name of section : Reagent division

Catalog and products information section

Address : 11-5 Nihonbashi Honcho 3-Chome Chuo-Ku,

Tokyo 103-0023 Japan

Telephone number : +81-3-3639-8301 Facsimile number : +81-3-3639-9435

MSDS No.49200-70 Date: 1998.06.03

Product name: COD meter A solution

Composition/Information on ingredients

Substance/Mixture: Substance

Chemical name : Potassium permanganate

Ingredients and composition: Potassium permanganate

about 0.08% water solution

Chemical formula : KMnO<sub>4</sub>
CAS registry number : 7722-64-7
UN class : Not applicable

Hazards Identification

Class name of hazardous chemicals for SDS in Japan : Not applicable

Physical and Chemical hazards:

This solution is noncombustible, and does not have

particular physical and chemical hazards.

Adverse human health hazards:

A large dose of swallow irritates to throat and stomach,

and causes nausea and vomiting.

Environmental effects: This substance may affect aquatic organisms.

First-aid measures

Eye contact : Gently rinse the affected eyes with clean water for

at least 15 minutes.

If necessary, get medical treatment.

Skin contact: Wash the affected areas under tepid running water.

Inhalation : Remove the victim from the contamination immediately to

fresh air. Keep them warm and quiet, and make them blow

their nose and gargle.

Ingestion : Rinse mouth with water. Give the victim one or two

glasses of water to dilute the chemical.

If necessary, get medical treatment.

Fire-fighting measures

The way fire-fighting:

This solution is noncombustible.

In case of fire, move container from fire areas if it can be done without risk. If it cannot be, apply water from a safe distance to cool and protect surrounding

area.

Firefighters should wear proper protective equipment.

Accidental release measures

Absorb spill with paper or cloths, then place in a chemical waste container.

Flush residual spill (area) with copious amounts of water.

Handling and storage

Handling: Avoid contact with skin or eyes.

If necessary, wear appropriate protective equipment.

Storage: Keep containers tightly closed, and store at a cool

place.

Exposure control/Personal protection

Control parameters ACGIH (1996): 5mg/m³ (as Manganese compounds)

Engineering measures: Use with local exhaust ventilation in vapor

atmospheres.

Personal protective equipment: If necessary, wear gloves and goggles.

Physical and chemical properties

Appearance : Purplish pink liquid, odorless

Boiling point : About 100
Melting point : About 0
Density : About 1

Solubility in water: Miscible in all proportion

Physical hazard

This solution is noncombustible.

Oxidizibility : Potassium permanganate itself has oxidizibilty, however,

0.08% water solution dose not have dangerous

oxidizibilty.

Stability and reactivity: Stable under normal usage.

Toxicological information

Corrosive property: If contact with skin, may cause irritation.

Acute toxicity : A large dose swallow irritates to throat and stomach,

and causes nausea and vomiting. (as potassium permanganate) rat oral LD $_{50}$ =1,090mg/kg rabbit oral LDL $_{0}$ =70mg/kg

Sub-chronic toxicity: Not available Chronic toxicity: Not available

Carcinogenic effects: Not listed on IARC or NTP.

Mutagenic effects: Not available

Effects on the reproductive system : Not available

<u>Teratogenic effects: Not available</u>

Ecological information

Biodegradability : Not available Bio-accumulation : Not available

Fish toxicity : LD<sub>50</sub>/96H=100-1mg/I(as Potassium permanganate)

Disposal consideration

Add reducing agent such as sodium sulfite solution

in this product to reduce.

After that, manganese oxide is precipitated.

Filter the precipitation and bury in the landfill site

approved for chemical and hazardous wastes.

Transport information

Follow all regulations in your country.

Regulatory information

Ensure this material in compliance with federal requirements and ensure conformity to local regulations.

References

Handbook of dangerous materials,

Guter Hommel, Springer-Verlag Tokyo (1991)

Chemical products of 13398, The Chemical Daily Co., Ltd (1998)

Chemical dictionary, Kyoritu publishing Co,.Ltd (1963)

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