

Nitrite Test

NO₂⁻

1. Method

In acidic solution nitrite ions react with sulfanilic acid to form a diazonium salt, which in turn reacts with N-(1-naphthyl)ethylenediamine dihydrochloride to form a red-violet azo dye. The nitrite concentration is measured **semiquantitatively** by visual comparison of the colour of the measurement solution with the colour fields of a colour disk.

2. Measuring range and number of determinations

Measuring range / colour-scale graduation ¹⁾	Number of determinations		
0.1 - 0.2 - 0.4 - 0.6 - 1.0 - 1.8 - 3.0 - 6.0 - 10 mg/l NO ₂ -	100		
0.03 - 0.06 - 0.12 - 0.18 - 0.30 - 0.55 - 0.9 - 1.8 - 3.0 mg/l NO ₂ -N	400		

1) for conversion factors see section 8

3. Applications

Sample material:

Groundwater, drinking water, and surface water Seawater Wastewater

Food after appropriate sample pretreatment Soils after appropriate sample pretreatment

4. Influence of foreign substances

This was checked in solutions containing 1 and 0 mg/l NO₂. The determination is not yet interfered with up to the concentrations of foreign substances given in the table.

Concentrations of foreign substances in mg/l or %							
Ca ²⁺ Cd ²⁺ CN ⁻ Cr ³⁺ Cr ₂ O ₇ ²⁻ Cu ²⁺ Fe ³⁺			100 1000 1000 1000 1000 1000 10 0			EDTA Reducing a (ascorbic a sulfite) NaCl NaNO ₃ Na ₂ SO ₄	

5. Reagents and auxiliaries

Please note the warnings on the packaging materials!

The test reagent is stable up to the date stated on the pack when stored closed at +15 to +25 $^\circ$ C.

Package contents:

- 3 bottles of reagent NO₂-1
- 1 graduated 6-ml plastic syringe
- 2 test tubes with screw caps

1 colour-disk comparator

Other reagents and accessories:

$$\label{eq:second} \begin{split} & \mathsf{Merckoquant}^{\circledast} \ \mathsf{Nitrite Test, Cat. No. 110007,} \\ & \mathsf{measuring range 2 - 80 \ \mathsf{mg/l NO_2}^{-} \ (0.6 - 24 \ \mathsf{mg/l NO_2}\text{-N}) \\ & \mathsf{Universal indicator strips pH 0 - 14, Cat. No. 109535} \\ & \mathsf{Acilit}^{\circledast} \ \mathsf{indicator strips pH 0 - 6.0, Cat. No. 109531} \\ & \mathsf{Sulfuric acid 0.5 \ \mathsf{mol/l TitriPUR}^{\circledast}, Cat. No. 109072} \\ & \mathsf{Sodium hydroxide solution 1 \ \mathsf{mol/l TitriPUR}^{\circledast}, Cat. No. 109137} \\ & \mathsf{Nitrite standard solution CertiPUR}^{\circledast}, 1000 \ \mathsf{mg/l NO_2}, Cat. No. 119899} \end{split}$$

Flat-bottomed tubes for Microquant® tests (12 pcs), Cat. No. 117988

Refill pack:

Cat. No. 118463 Nitrite Test

Refill pack for Microquant[®] 114774, Aquaquant[®] 114424, and Aquaquant[®] 114408

(Reagent **without technical accessories** for the number of determinations stated in section 2)

6. Preparation

- Analyze immediately after sampling.
- Check the nitrite content with the Merckoquant[®] Nitrite Test. Samples containing more than 10 mg/l NO₂⁻ must be diluted with distilled water.
- The pH must be within the range 2 10. Adjust, if necessary, with sulfuric acid.
- Filter strongly turbid samples.

	Measurement sample right-hand tube (A) behind the colour disk	Blank left-hand tube (B) behind the colour disk		
Pretreated sample (15 - 25 °C)	6 ml	6 ml	Inject into the test tube with the syringe.	
Reagent NO ₂ -1	1 level blue microspoon (in the cap of the NO ₂ -1 bottle)	-	Add, close the tube, and shake vigorously until the reagent is completely dissolved. The pH must be within the range 2.0 - 2.5. Check with Acilit [®] indicator strips. Adjust the pH, if necessary, with sodium hydroxide solu- tion or sulfuric acid.	
Leave to stand for 3 min (reaction time).				

Notes on the measurement:

termediate value.

7. Procedure

• The colour of the measurement solution remains stable for at least 60 min after the end of the reaction time stated above.

Read off the result in mg/l NO2⁻ shown in the small window or, if necessary, estimate an in-

- Turbidity in the measurement solution makes the colour comparison more difficult.
- If the colour of the measurement solution is equal to or more intense than the darkest colour on the scale, repeat the measurement using **fresh**, diluted samples until a value of less than 10 mg/l NO₂⁻ is obtained.

Concerning the result of the analysis, the dilution (see also section 6) must be taken into account:

Result of analysis = measurement value x dilution factor

8. Conversions

Units required	= units given	x conversion factor
mg/I NO ₂ -N	mg/l NO ₂ -	0.304
mg/I NO ₂ -	mg/I NO₂- N	3.28

9. Method control

To check test reagent, measurement device, and handling: Dilute the nitrite standard solution with distilled water to 1.0 mg/l NO_2 and analyze as described in section 7.

Additional notes see under www.merck-chemicals.com/qa.

10. Notes

- Reclose the reagent bottle immediately after use.
- Rinse the test tubes and the syringe with distilled water only.