MQuant™

Total Hardness Test

7 75306 0002-xxxxxxxxxx msn

1. Definition

The hardness (total hardness) of a given water is due to its content of salts of the alkaline earth metdue to its content of saits of the arkaline earth mer-als calcium, magnesium, strontium, and barium ("hardening constituents"). Since strontium and bar-ium are generally present in waters only in traces, the hardness is defined as the content in a water of calcium ions, Ca²⁺, and magnesium ions, Mg²⁺ ("hard-ness ions"). The conventional procedure is to relate the statement of the water hardness only to calcium, in other words to express also the content of magnesium ions as calcium content.

The units for the water hardness relate to calcium or its compounds CaO (1 °d \triangleq 10 mg/l CaO) or CaCO₃ (1 °e \triangleq 14.25 mg/l CaCO₃; 1 °f \triangleq 10 mg/l CaCO₃), with the magnesium content being expressed as calcium content and included in the calculation accordingly.

2. Method

In the presence of a green indicator the hardness ions Ca2+ und Mg2+ react with Titriplex® III 1) to form a colorless, stable complex. The hardness ions that are present in excess relative to Titriplex® III react with the indicator to form a red complex. The reaction zones of the test strip, which are impreg nated with differing amounts of Titriplex® III, thus assume a green to red color, depending on the hardness of the water sample. The hardness is measured semiquantitatively by visual comparison of the reaction zones with the color rows of a color scale.

1) ethylenedinitrilotetraacetic acid disodium salt dihydrate

3. Measuring range and number of determinations

Measuring range / color- scale gradation 1)	Number of determinations
<150; 150 - 250; >250 mg/l CaCO ₃	
<1.5; 1.5 - 2.5; >2.5 mmol/l CaCO ₃ <10.5; 10.5 - 17.5; >17.5 °e	100

¹⁾ for conversion factors see section 9

4. Applications

Sample material:

Groundwater and surface water Drinking water Mineral water

5. Influence of foreign substances

This was checked in solutions with a hardness of 0 mmol/l $CaCO_3$. The determination is not yet interfered with up to the concentrations of foreign substances given in the table.

Concentrations of foreign substances in mg/l						
Cu ²⁺ Co ²⁺	10	Ni ²⁺ Zn ²⁺	10			

6. Reagents and auxiliaries

The test strips are stable up to the date stated on the pack when stored closed at +15 to +25 °C.

Package contents:

Tube containing 100 test strips

Other reagents:

MColorpHast™ Universal indicator strips pH 0 - 14, Cat. No. 109535

Sodium hydroxide solution 1 mol/l TitriPUR®, Cat. No. 109137

Hydrochloric acid 1 mol/l TitriPUR®,

Cat. No. 109057

Calcium chloride dihydrate for analysis EMSURE®,

Cat. No. 102382

7. Preparation

- The pH must be within the range 5 8. Adjust, if necessary, with sodium hydroxide solution or hydrochloric acid.
- Other MQuant™ Total Hardness Tests for measuring range <4 - >26 °e: Cat. No. 110025 - 100 test strips for measuring range >6 - >31 °e: Cat. No. 110046 - 100 test strips

8. Procedure

Immerse both reaction zones of the test strip in the pre-treated sample (15 - 30 °C) for 1 sec (not in running

Shake off excess liquid from the strip and after 1 min determine with which color row on the label the colors of the reaction zones coincide most exactly.

Read off the corresponding result.

Assessment:

Hardness range	mg/l CaCO₃	mmol/l CaCO ₃ (Ca)	°e
soft	<150	<1.5	<10.5
moderately hard	150 - 250	1.5 - 2.5	10.5 - 17.5
hard	>250	>2.5	>17.5

Note on the measurement:

The color of the reaction zones may continue to change after the specified reaction time has elapsed. This must not be considered in the measurement

9. Conversions

required given	mmol/l CaCO ₃ (Ca)	mg/l CaCO₃	mg/I Ca	English degree °e	French degree °f	German degree °d
1 mmol/l CaCO ₃ (Ca)	1	100.1	40.08	7.02	10.01	5.61
1 mg/l CaCO ₃	0.010	1	0.400	0.070	0.100	0.056
1 mg/l Ca	0.025	2.50	1	0.175	0.250	0.140
1 English degree °e	0.142	14.25	5.71	1	1.43	0.799
1 French degree °f	0.100	10.00	4.00	0.702	1	0.560
1 German degree °d	0.178	17.85	7.15	1.25	1.78	1

10. Method control

To check test strips and handling:

Dissolve 3.67 g of calcium chloride dihydrate in distilled water, make up to 1000 ml with distilled water, and mix. Ca content: 1000 mg/l (≜ 25 mmol/l CaCO₃). Dilute this standard solution with distilled water to 80 mg/l Ca (≙ 2.0 mmol/l CaCO₃) and analyze as described in section 8.

Additional notes see under www.qa-test-kits.com.

Note

Reclose the tube containing the test strips immediately after use.

Merck KGaA, 64271 Darmstadt, Germany, Tel. +49(0)6151 72-2440 www.analytical-test-kits.com

EMD Millipore Corporation, 290 Concord Road, Billerica, MA 01821, USA, Tel. +1-978-715-1335

