Spectroquant®

Ammonium Cell Test



1. Method

Ammonium nitrogen (NH₄-N) occurs partly in the form of ammonium ions and partly as ammonia. A pH-dependent equilibrium exists between the two forms. In strongly alkaline solution ammonium nitrogen is present almost entirely as ammonia, which reacts with hypochlorite ions to form monochloramine. This in turn reacts with a substituted phenol to form a blue indophenol derivative that is determined photometrically.

The method is analogous to EPA 350.1, APHA 4500-NH₃ F, ISO 7150-1, and DIN 38406-5.

2. Measuring range and number of determinations

Measuring range	Number of determinations
4.0 - 80.0 mg/l NH ₄ -N	05
5.2 - 103.0 mg/l NH ₄ +	25

For programming data for selected photometers / spectrophotometers see www.service-test-kits.com.

3. Applications

This test measures both ammonium ions and dissolved ammonia.

Sample material:

Groundwater and surface water, seawater

Drinking water

Wastewater

Nutrient solutions for fertilization

Soils and food after appropriate sample pretreatment

4. Influence of foreign substances

This was checked in solutions containing 40 and 0 mg/l NH₄-N. The determination is not yet interfered with up to the concentrations of foreign substances given in the table.

	Concentrations of foreign substances in mg/l or %					
Al ³⁺	1000	Mn ²⁺	100	EDTA	1000	
Ca ²⁺	1000	Ni ²⁺	250	Primary amines ¹⁾	0	
Cd ²⁺	1000	NO ₂	1000	Secondary amines ²⁾	100	
CN-	250	Pb ²⁺	1000	Aminophenols	100	
Cr ³⁺	100	PO ₄ 3-	1000	Aniline	250	
Cr ₂ O ₇ ²⁻	1000	S ²⁻	50	Triethanolamine	1000	
Cu ²⁺	250	SiO ₃ ²⁻	1000	Surfactants ³⁾	1000	
F·	1000	Zn ²⁺	500	Na-acetate	10 %	
Fe ³⁺	250			NaCl	20 %	
Hg ²⁺ Mg ²⁺	500			NaNO₃	20 %	
Mg ²⁺	500			Na₂SO₄	20 %	

Reducing agents interfere with the determination

- 1) tested with methylamine
- 2) tested with dimethylamine
- 3) tested with nonionic, cationic, and anionic surfactants

5. Reagents and auxiliaries

Please note the warnings on the packaging materials!

The test reagents are stable up to the date stated on the pack when stored closed at +15 to +25 °C.

Package contents:

- 1 bottle of reagent NH₄-1K (contains granulate + desiccant capsule) 25 reaction cells
- 1 blue dose-metering cap
- 1 sheet of round stickers for numbering the cells

Other reagents and accessories: MQuant™ Ammonium Test, Cat. No. 110024,

measuring range 10 - 400 mg/l NH $_4$ * (8 - 311 mg/l NH $_4$ -N) MColorpHastTM Universal indicator strips pH 0 - 14, Cat. No. 109535

Sodium hydroxide solution 1 mol/l TitriPUR®, Cat. No. 109137 Sulfuric acid 0.5 mol/l TitriPUR®, Cat. No. 109072

Spectroquant® CombiCheck 70, Cat. No. 114689

Ammonium standard solution CRM, 6.00 mg/l NH₄-N, Cat. No. 125025

Ammonium standard solution CRM, 12.0 mg/l NH₄-N, Cat. No. 125026

Ammonium standard solution CRM, 50.0 mg/l NH₄-N, Cat. No. 125027

Pipette for a pipetting volume of 0.10 ml

6. Preparation

• Rinse glassware ammonium-free with distilled water. Do not use detergent!

At the first use replace the screw cap of the reagent bottle NH₄-1K by the blue dose-metering cap

Hold the reagent bottle vertically and, at each dosage, press the slide all the way into the dose-metering cap. Before each dosage ensure that the slide is completely retracted.



Reclose the reagent bottle with the screw cap at the end of the measurement series, since the function of the reagent is impaired by the absorption of atmospheric moisture.

- Analyze immediately after sampling.
- Check the ammonium content with the MQuant™ Ammonium Test. Samples containing more than 80.0 mg/l NH₄-N must be diluted with distilled water.
- The pH must be within the range 4 13.

Adjust, if necessary, with sodium hydroxide solution or sulfuric acid.

Filter turbid samples.

7. Procedure

Pretreated sample (20 - 30 °C)	0.10 ml	Pipette into a reaction cell (20 - 30 °C), close the cell, and mix.
Reagent NH ₄ -1K	1 dose	Add, close the cell tightly, and shake vigorously until the reagent is completely dissolved.

Leave to stand for 15 min (reaction time), then measure the sample in the photometer.

Notes on the measurement:

- For photometric measurement the cells must be clean. Wipe, if necessary, with a clean dry cloth.
- Measurement of turbid solutions yields false-high readings.
- Ammonium-free samples turn yellow on addition of reagent NH₄-1K.
- The pH of the measurement solution must be within the range 11.5 11.8.
- The color of the measurement solution remains stable for at least 60 min after the end of the reaction time stated above.
- In the event of ammonium concentrations exceeding 5000 mg/l, other reaction products are formed and false-low readings are yielded. In such cases it is advisable to conduct a plausibility check of the measurement results by diluting the sample (1:10, 1:100).

8. Analytical quality assurance

recommended before each measurement series

To check the photometric measurement system (test reagent, measurement device, handling) and the mode of working, the ammonium standard solutions CRM, 6.00 mg/l NH $_4$ -N, Cat. No. 125025, 12.0 mg/l NH $_4$ -N, Cat. No. 125026, and $50.0~mg/l~NH_4-N,~Cat.~No.~125027~or~Spectroquant <math display="inline">^{\circ}$ CombiCheck 70 can be used. Besides a standard~solution with $50.0~mg/l~NH_4-N,~CombiCheck~70~also~constant <math display="inline">^{\circ}$ tains an addition solution for determining sample-dependent interferences (matrix effects).

Additional notes see under www.qa-test-kits.com.

Characteristic quality data:

In the production control, the following data were determined in accordance with ISO 8466-1 and DIN 38402 A51:

Standard deviation of the method (mg/l NH ₄ -N)	± 0.48
Coefficient of variation of the method (%)	± 1.2
Confidence interval (mg/l NH ₄ -N)	± 1.2
Number of lots	42

Characteristic data of the procedure:

Sensitivity: Absorbance 0.010 A corresponds to (mg/l NH ₄ -N)	0.4
Accuracy of a measurement value (mg/l NH_4 -N)	max. ± 1.9

For quality and batch certificates for Spectroquant® test kits see the website.

9. Notes

- Reclose the reagent bottle immediately after use.
- Information on disposal can be obtained at www.disposal-test-kits.com.

