Name of manufacturer: KANTO CHENICAL CO., INC.

Name of section : Reagent division

Catalog and products information section

: 11-5 Nihombasi Honcho 3-Chome Chuo-ku. Address

Tokyo 103-0023 Japan

Telephone number

: (03)3639-8301

Facsimile number

: (03)3639~9435

MSDS No.

Date: 1998. 6. 3

determination

Composition/Information on ingredients

A

Substance/Mixture: Substance

Potassium permanganate Chemical name

Ingredients and composition: Potassium permanganate

about 0.08% water solution

Chemical formula: CAS registry number : 7722-64-7

KMnO 4

UN class

Product mame:

Not applicable

Hazards Identification

Class name of hazardous chemicals for SDS in Japan: Not applicable Physical and Chemical hazards :

solution for COD

This solution is noncombustible, and does not have

particular physical and chemical hazards.

Adverse human health hazards:

A large dose of swallow irritates to throat, stomach and causes nausea, vomiting.

Environmental effects: This substance may affect aquatic organisms.

First-aid measures

Gently rinse the affected eyes with clean water at least Eye contact :

15 minutes.

If necessary, get medical treatment.

Remove all contaminated clothing, shoes and socks from Skin contact:

the affected areas as quickly as possible. Wash the

affected areas under tepid running water.

Inhalation : Remove the victim from the contamination immediately to

fresh air. Keep them warm and quiet, and make them blow

their nose and gargle.

Ingestion : Rinse mouth with water. Give the victim one or two

glasses of water to dilute the chemical.

If necessary, get medical treatment.

Fire-fighting measures

The way fire-fighting :

This solution is noncombustible.

In case of fire, move container from fire areas if it can be done without risk. If it cannot be, apply water from a safe distance to cool and protect surrounding area.

Firefighters should wear proper protective equipment.

Accidental release measures

Absorb spill with paper or cloths, then place in a

chemical waste container.

Flush residual spill (area) with copious amounts of

water.

Handling and storage

Handling: Avoid contact with skin or eyes,

If necessary, wear appropriate protective equipment.

Storage : Keep containers tightly closed, and store at a cool

place.

Exposure control/Personal protection

Control parameters ACCIH (1996): 5mg/m 3 (as Manganese compounds) Engineering measures: Use with local exhaust ventilation in vapor

atmospheres.

Personal protective equipment: If necessary, wear gloves, goggles.

Physical and chemical properties

Apparance : Violescer

: Violescent red liquid, odorless

Boiling point : About 100 T Welting point : About 0 T Density : About 1

Solubility in water: Miscible in all proport

Physical hazard

This solution is noncombustible.

Oxidizibilty

Potassium permangante itself has oxidizibilty, however,

0.08% water solution dose not have dangerous

oxidizibilty.

Stability and reactivity: Stable under normal usage.

Toxicological information

Corrosive property: If contact with skin, may cause irritation.

Acute toxicity : A large dose swallow irritates to throat, stomach

and causes nausea, vomiting. (as potassium permanganate) rat oral LD₅₀=1,090mg/kg rabbit oral LDL o =70mg/kg

Sub-chronic toxicity: Not available Chronic toxicity: Not available Carcinogenic effects: Not available Mutagenic effects: Not available

Effects on the reproductive system : Not available

Teratogenic effects: Not available

Ecological information

Biodegradablity : Not available Bioaccumulation : Not available

Fish toxicity : LD₅₀/96H=100-1mg/1(as Potassium permanganate)

Disposal consideration

Add reducing agent such as sodium sulfite solution in this product to reduce.

After that, manganese oxide is deposited.

Filter the deposit and bury in the landfill site

approved for chemical and hazardous wastes.

Transport information

Follow all regulations in your country.

Regulatory information

Ensure this material in compliace with federal requirements and ensure comformity to local regulations.

Refferences

DHandbook of dangerous materials,

Giter Hommel. Springer-Verlag Tokyo (1991)

- @Chemical products of 13398. The Chemical Daily Co., Ltd(1998)
- 3 Chemical dictionary, Kyoritu publishing Co., Ltd (1963)
- Dangerous Properties of Industrial Materials, 6th ed N. I. Sax Van Nostrand Reinhold Company (1984)
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